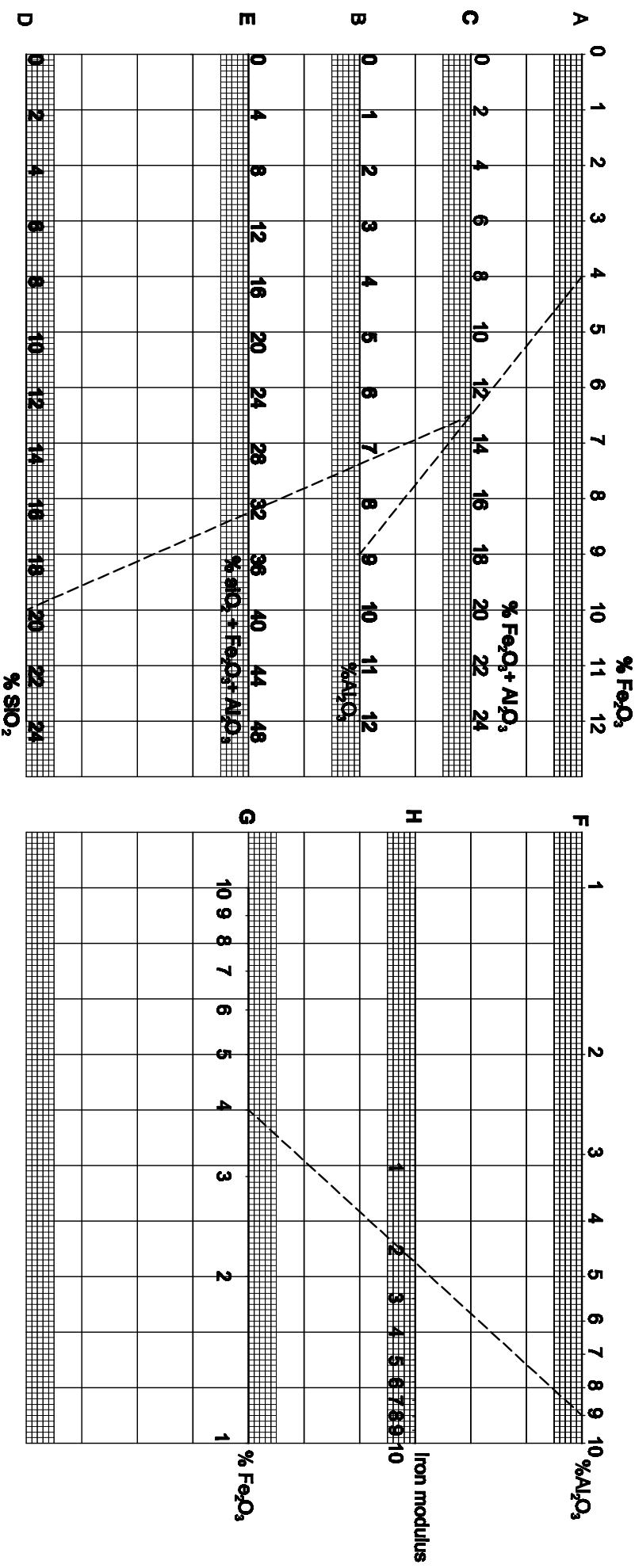


step 1

calculate $\text{Fe}_2\text{O}_3 + \text{Al}_2\text{O}_3$ & $\text{Fe}_2\text{O}_3 + \text{Al}_2\text{O}_3 + \text{SiO}_2$



scale :

A : 1 cm = 1%, B : 1 cm = 1%, C : 1 cm = 2%
D : 1 cm = 2%, E : 1 cm = 4%

example:

$\text{Fe}_2\text{O}_3 = 4\%$ (A), $\text{Al}_2\text{O}_3 = 9\%$ (B), $\text{Fe}_2\text{O}_3 + \text{Al}_2\text{O}_3 = 13\%$ (C)
 $\text{Fe}_2\text{O}_3 + \text{Al}_2\text{O}_3 = 13\%$ (C) $\text{SiO}_2 = 20\%$ (D)
 $\text{Fe}_2\text{O}_3 + \text{Al}_2\text{O}_3 + \text{SiO}_2 = 33\%$ (E)

step 2 calculate Iron modulus $\text{Al}_2\text{O}_3/\text{Fe}_2\text{O}_3$

scale :

F & G log scale 10 cm = 1-10 %
H = log scale 5 cm = 1-10 ratio

example:

$\text{Al}_2\text{O}_3 = 9\%$ (F)
 $\text{Fe}_2\text{O}_3 = 4$ (G)
 $\text{Al}_2\text{O}_3 / \text{Fe}_2\text{O}_3 = \text{Iron modulus} = 2.25$ (H)